

## B.A. Sophomore Year – Spring Semester Lab

### CEM 255 Organic Laboratory

**Description:** Preparation and qualitative analysis of organic compounds.

**Credit:** 2 Credits (3 hours lab, 1 hour of lecture per week)

**Prerequisite:** CEM 252 or concurrent enrollment in CEM 252

Students are introduced to microscale techniques in organic chemistry laboratory. They also perform a few synthetic experiments including a multi-step synthesis. There is a hands-on-experiment with  $^1\text{H}$ NMR (300MHz). The lab reports are informal and they include purpose, procedure, results, and discussion.

#### Experiments:

##### 1. Calibration of Thermometer and Molecular Model Experiment

Students calibrate their thermometer and do a molecular model experiment during the check in.

##### 2. Thin Layer Chromatography (TLC)

Students learn TLC technique and identify a two-component mixture by using the six standards.

##### 3. Distillation

a. Students perform simple and fractional distillation, then calculate the number of theoretical plate (by injecting their products into a GC) for each of the experiments and compare the efficiency of their distillation set-ups.

b. Synthesis of cyclohexene from cyclohexanol. Fractional distillation is used to remove cyclohexene as it is made in order to achieve quantitative yields.

##### 4. Extraction

Microscale technique is used to separate a carboxylic acid from an ester by performing an acid–base extraction.

##### 5. Identification of an unknown using NMR–Spectroscopy

Students are to identify the identity of a liquid unknown sample of an aldehyde or ketone by taking the boiling point (microscale), making a solid derivative, and taking NMR spectra and fully characterizing it.

##### 6. Multistep Synthesis (Semimicro)

The conversion of methyl benzoate into 3–nitrobenzamide in a three step synthesis.

##### 7. Microscale Grignard Reaction

This experiment is a microscale reaction of phenyl magnesium bromide (Grignard reagent) with methyl benzoate.

##### 8. Azo Dye Synthesis (Semimicro)

Preparation of Orange II dye by forming the diazonium salt of sulfanilic acid and coupling it with sodium salt of  $\beta$ -naphthol.